News Release Mar 6, 2018 GS Yuasa Corporation



# GS Yuasa Succeeds in Using Silicon-metal Electrode Development to Improve the High Energy Density Technology in Lithium-ion Batteries

GS Yuasa Corporation (Tokyo Stock Exchange: 6674; "GS Yuasa") announced that it has both developed high energy density technology as well as longer lifecycles for silicon metal-based negative electrodes. Both of these properties have posed many challenges with respect to the commercialization of large batteries. As a result, the energy density of batteries, the size of which are used in actual electric vehicles, has been successfully increased by about three times compared to that of traditional lithium-ion batteries.

The silicon metal used in the negative electrode material has an extremely high theoretical capacity (4200 mAh/g), and there is an abundance of silicon available. Therefore, much research has been conducted on silicon metal as a new material for lithium-ion batteries. However, since silicon metal undergoes an extremely large change in volume of approximately 400% in conjunction with the charge-discharge cycle, degradation including pulverization<sup>\*1</sup> and isolation<sup>\*2</sup> occurs as the battery is repeatedly charged and discharged. As a result, such batteries have low coulombic efficiency and poor cycling performance, and therefore practical application of silicon metal electrodes has been considered a difficult challenge, particularly for large batteries used in electrically powered vehicles, which must perform for long periods of time. To address this challenge, GS Yuasa has succeeded in improving coulombic efficiency and cycling performance by finding the optimal particle size and electrode structure of electrodes using silicon metal, thereby improving high-energy density technology to a level roughly three times greater than up until now.

This silicon metal electrode is technology that can be applied to all-solid-state batteries, for which future technological innovation and more widespread use are expected going forward.

Going forward, GS Yuasa will further improve the cycling performance of this silicon metal electrode, with the aim of applying it in electrically powered vehicles by around 2025.

GS Yuasa will continue to contribute to the realization of a low-carbon society through the development of high energy density technology for batteries, which is necessary for the electrification of vehicles.

[Highlights of Technology Improvements]

## 1. Higher initial coulombic efficiency and cycling performance improvement by selecting optimal particle size of silicon metal

When the particle size is small, the initial coulombic efficiency is low, and when the particle size is large, cycling performance is weak due to the fact that pulverization becomes prominent. GS Yuasa found that applying silicon metal with an optimal particle size improves these two properties.

## 2. Higher discharge properties by combining various conductive additives

Normally, a single type of conductive additive is used in an electrode. However, in the case of a silicon metal electrode, the electrode formability improved by using multiple conductive additives. Furthermore, it became clear that there was a higher level of discharge due to the improved conductivity of the electrode.

### 3. Improved suitability for mass production due to the application of a water-soluble binder

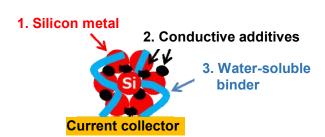
In order to produce silicon metal electrodes, processes not suitable for mass production sometimes arise, such as adding high temperature heat treatment in an inert atmosphere\*3, but by applying a water-soluble binder, the process has been simplified. As a result, we succeeded in finding the electrode structure which is highly-suitable for mass production while still maintaining a high level of binding strength.

- \*1 A phenomenon in which strains caused by the expansion and contraction of a particle develop into cracks, and the particle is broken into smaller.
- \*2 A phenomenon in which the active material in the electrode stops contributing to charging and discharging reaction due to electrolyte degradation, electrode expansion, etc.
- \*3 A vacuum state, a state of reduced pressure, or a state in which there are many inactive gases such as nitrogen and argon in the surroundings.

#### <Notice>

Some of these results will be presented at the 85<sup>th</sup> ECSJ (Electrochemical Society of Japan) Spring Meeting (Sponsored by the Electrochemical Society of Japan) held March 9-11 at Tokyo University of Science's Katsushika Campus. http://www.electrochem.jp/program/2018spring/2018spring.html

[Schematic Diagram of Silicon Metal Electrode]





[Charge-discharge curves for high capacity lithium-ion battery using silicon metal]

